## Revealing the Link between Structural Relaxation and Dynamic Heterogeneity in Glass-Forming Liquids

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Despite the use of glasses for thousands of years, the nature of the glass transition is still mysterious. On approaching the glass transition, the growth of dynamic heterogeneity has long been thought to play a key role in explaining the abrupt slowdown of structural relaxation. However, it still remains elusive whether there is an underlying link between structural relaxation and dynamic heterogeneity. Here, we unravel the link by introducing a characteristic time scale hiding behind an identical dynamic heterogeneity for various model glass-forming liquids. We find that the time scale corresponds to the kinetic fragility of liquids. Moreover, it leads to scaling collapse of both the structural relaxation time and dynamic heterogeneity for all liquids studied, together with a characteristic temperature associated with the same dynamic heterogeneity. Our findings imply that studying the glass transition from the viewpoint of dynamic heterogeneity is more informative than expected.

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Nowadays, various theoretical and empirical equations, e.g., mode coupling (MC), Vogel-Fulcher-Tammann (VFT), Elmatad-Chandler-Garrahan (ECG), Avramov-Milchev (AM), and Mauro-Yue-Ellison-Gupta-Allan (MYEGA) forms [1-6], are proposed to fit the structural relaxation time of glass-forming liquids  $\tau(T)$ , as a function of temperature T, and to interpret the glass transition in different theoretical frameworks. Despite the diversity of the fitting functions, the kinetic fragility [7],  $m = \partial(\log \tau)/$  $\partial (T_q/T)|_{T=T_q}$ , with  $T_q$  being the glass transition temperature, is commonly employed to evaluate the deviation of  $\tau(T)$  from the Arrhenius behavior [1], which proposes a useful classification of liquids along a "strong" to "fragile" scale [7]. Thus, the scaling collapse of discrete  $\tau(T)$  data in various glass-forming liquids is believed to be an effective way to simplify the elusive glass transition [1-3]. Although great efforts have been devoted [8-11], it is still unclear whether there is a general and simple description (without introducing adjustable free parameters) of  $\tau(T)$  for glassforming liquids with vastly different m.

In the past decades, one grail in the study of glasses is the finding of dynamic heterogeneity referring to the spatiotemporal fluctuations in local dynamics [12–14]. The growth of the dynamic heterogeneity and its dynamic correlation length [15–17] as T decreases toward the glass transition provides a possible approach to understand the dramatic slowdown of dynamics during vitrification. Thus, more attention [18–28] has been attracted to investigate the correlation between structural relaxation and dynamic heterogeneity in glass-forming liquids. The critical issue nowadays is that experimental and numerical studies [19–24] have showed that dynamic heterogeneities in state points under isochronal condition (i.e., constant  $\tau$ ) can be either invariant or variant. Recently, attempts have been made to search for the general relation between structural relaxation and dynamic heterogeneity [18,26,27], but there seems to be no consensus on it [20,24,25,28]. Furthermore, the concept of fragility is believed to be correlated well with dynamic heterogeneity in model glass-forming liquids [29,30], while an experimental study [15] reported that there was no convincing correlation between them. To our knowledge, even in model glass formers where a precise quantification of dynamic heterogeneity is feasible, the directly quantitative evidence for the correlation between fragility and dynamic heterogeneity is still lacking.

In this Letter, we reveal the underlying connection between dynamic heterogeneity and structural relaxation by introducing a characteristic time scale hidden in state points with an identical dynamic heterogeneity in different model glassforming liquids. This time scale corresponds to the kinetic fragility of glass-forming liquids and bridges structural relaxation and dynamic heterogeneity by achieving fantastic scaling collapses. Moreover, a rather general description of  $\tau(T)$  for various glass-forming liquids can be achieved from the viewpoint of a constant dynamic heterogeneity condition without introducing any free parameter.

We perform extensive molecular dynamics simulations in the *NPT* (constant number of particles N, pressure P, and temperature T) ensemble in six potential models [31]:

harmonic (Harm), Hertzian (Hertz), 12-6 repulsive Lennard-Jones (RLJ), 36-6 RLJ, 12-6 Lennard-Jones (LJ), and embedded atom method (EAM) potentials. We measure the self-part of the intermediate scattering function [35]  $F_s(k,t) = \frac{1}{N} \langle \sum_{j=1}^N \exp\{i\vec{k} \cdot [\vec{r}_j(t) - \vec{r}_j(0)]\} \rangle$ , where  $\vec{r}_i(t)$  is the location of particle *j* at time *t*,  $|\vec{k}|$  takes approximately the value at the first peak of the static structure factor [36], and  $\langle . \rangle$  denotes the time average. The structural relaxation time  $\tau$  is defined by the relation [31]  $F_s(k,\tau) = e^{-1}$ . Dynamic heterogeneity is quantified by the time-dependent non-Gaussian parameter [12,25,37]  $\alpha_2(t) = \frac{3}{5} \langle \Delta r^4 \rangle / \langle \Delta r^2 \rangle^2 - 1$ , where  $\Delta r$  is the displacement of a particle during time t. In the Supplemental Material [31], we also show results regarding dynamic heterogeneity characterized by the four-point dynamic susceptibility  $\chi_4$ [17,23,38].

Figure 1(a) shows the Angell plots of  $\tau$  versus  $T_{\rm ref}/T$  for six glass-forming systems with different potentials and pressures.  $T_{\rm ref}$  is a reference temperature at which  $\tau = \tau_q$ is sufficiently large in the endurable time window of simulation and identical for all systems, which is treated here as  $T_a$  to calculate the kinetic fragility *m*. Two systems have identical m if their curves in Fig. 1(a) coincide, and a steeper curve represents a more fragile liquid with a larger m. Systems with different potentials could exhibit the same *m*, as illustrated by the collapse of curves with harmonic and Hertzian potentials at the same pressures. With increasing pressure, the kinetic fragility increases, consistent with previous simulation studies [29,39,40], whereas the pressure dependence of fragility in most real liquids [41-43] is different from model ones. Therefore, by varying the pressures and potentials, we are able to investigate systems with vastly different values of m [31] (the range of m is still not as large as that in real materials [41–43]).

On approaching the glass transition,  $\alpha_2(t)$  exhibits a nonmonotonic *t* dependence with a maximum  $\alpha_{2,\text{max}}$  occurring at  $t = \tau_{\alpha_{2,\text{max}}}$  (see examples in Fig. S2 in the Supplemental Material [31]). As expected [12,25,37], both  $\alpha_{2,\text{max}}$  and  $\tau_{\alpha_{2,\text{max}}}$  increase when *T* decreases. Figure 1(b)



FIG. 1. (a) Angell plots of structural relaxation time  $\tau$  versus scaled reciprocal temperature  $T_{\rm ref}/T$  for systems with different potentials and pressures.  $T_{\rm ref}$  is determined according to  $\tau(T_{\rm ref}) = \tau_g \approx 2.16 \times 10^4$ . (b) Correlation between  $\tau$  and  $\alpha_{2,\rm max}$ . Symbols in (a) and (b) have the same meanings.

shows the correlation between  $\alpha_{2,\text{max}}$  and  $\tau$  for the same systems shown in Fig. 1(a). For each system,  $\alpha_{2,\text{max}}$ increases with increasing  $\tau$ , indicating that dynamic heterogeneity grows with the slowdown of structural relaxation during vitrification [14–17]. Along with Figs. 1(a) and 1(b) shows that, under the isochronal condition, one system with a larger *m* exhibits a larger  $\alpha_{2,\text{max}}$ . Therefore, more fragile liquids are more heterogeneous in dynamics [29,30]. More importantly, systems with the same *m* also exhibit identical  $\alpha_{2,\text{max}}(\tau)$ , which implies that the kinetic fragility is very likely to be the long-sought key parameter to connect structural relaxation and dynamic heterogeneity.

Figure 2(a) shows that we can collapse the  $\alpha_{2,\max}(\tau)$  curves for all systems investigated onto a single master curve when  $\tau$  is scaled by  $\tau^*$ , and hence

$$\alpha_{2,\max} = f_{\tau}(\tau/\tau^*), \tag{1}$$

where  $f_{\tau}$  is piecewise [see fitting lines in Fig. 2(a)]. Here, the scaling parameter  $\tau^*$  is a system-dependent characteristic time scale for all systems to have the same  $\alpha_{2,\text{max}}$ , i.e., under the iso- $\alpha_{2,\text{max}}$  condition. We choose a Hertzian system at  $T = 1.46 \times 10^{-4}$  and  $P = 5.00 \times 10^{-3}$  as a reference state, for which  $\tau^* \approx 3.25 \times 10^3$  and  $\alpha_{2,\text{max}} \approx 1.67$ . The scaling collapse is obtained by shifting all other curves



FIG. 2. (a) Maximum non-Gaussian parameter  $\alpha_{2,\text{max}}$  versus reduced structural relaxation time  $\tau/\tau^*$ , with  $\tau^*$  being the characteristic time scale under the iso- $\alpha_{2,\text{max}}$  condition  $(\alpha_{2,\text{max}} \approx 1.67 \text{ here})$ . The solid lines are fits to  $\alpha_{2,\text{max}} \sim (\tau/\tau^*)^{\nu}$ , with  $\nu = 0.8$  (black line) and  $\nu = 0.3$  (red line). (b) Correlation between  $\tau^*$  and kinetic fragility *m*, where *m* is calculated at  $T_g = T_{\text{ref}}$  as set in Fig. 1(a). The black solid line is a fit to  $\tau^* \sim m^{-\gamma}$ , where  $\gamma = 3.3$ . (c) Universal scaling between  $\tau$  and  $\tau_{\alpha_{2,\text{max}}}$ . The solid lines are fits to  $\tau/\tau^* \sim (\tau_{\alpha_{2,\text{max}}}/\tau^*)^{\beta}$ , where  $\beta = 1.2$  (black line) and  $\beta = 1.5$  (red line). (d)  $\alpha_{2,\text{max}}$  versus scaled reciprocal temperature  $T^*/T$ , with  $T^*$  being the characteristic temperature. The black and red solid lines are fitting curves consistent with the VFT fitting in Fig. 3 and power-law fittings in Fig. 2(a) [Eq. (4) can be derived from Eqs. (1) and (5)]:  $\alpha_{2,\text{max}} = 0.093 \exp[1.285/(T/T^* - 0.705)]$  and  $\alpha_{2,\text{max}} = 0.299 \exp[0.482/(T/T^* - 0.705)]$ , respectively.

onto that of the reference state. Surprisingly, Fig. 2(b) shows that  $\tau^* \sim m^{-\gamma}$ , so Eq. (1) can be rewritten as

$$\alpha_{2,\max} = f_m(\tau m^\gamma),\tag{2}$$

where  $\gamma$  varies with the time  $\tau_g$  used to evaluate *m*. As shown in Figs. 1(a) and 2(b), here we choose  $\tau_q \approx 2.16 \times 10^4$  and  $\gamma \approx 3.3$  and *m* takes values from 2.5 to 32.4 (in Fig. S3 of the Supplemental Material [31], we show another example with  $\gamma \approx 4.5$  and  $m \in [7.9, 54.6]$ , when  $\tau_q = 10^6$  is chosen). Recent studies [19-24] showed that whether dynamic heterogeneities at constant  $\tau$  vary depends on different control parameters, e.g., the softness of atomic interactions [19], pressures [20,22], and density scaling [23,24]. However, to our knowledge, it remains unknown whether there is a single parameter that controls the correlation between dynamical heterogeneity and  $\tau$  in different glass formers. Our findings in Figs. 1 and 2 suggest that it is the kinetic fragility that couples with the characteristic time scale hiding behind the iso- $\alpha_{2,max}$  condition and plays a key role in establishing the general relation between  $\alpha_{2,\max}$  and  $\tau$ . Thus, Eq. (2) reveals quantitatively the underlying correlation between dynamic heterogeneity and structural relaxation in glass-forming liquids.

It is interesting to compare experimental results with ours. Some experimental studies [15,44] reported no convincing (or weak) correlation between fragility and dynamic heterogeneity from an indirect measure of  $\chi_4$ , which is contrary to the obvious correlation shown in Fig. 2(b). Further studies are thus required to resolve the disagreement between experiments of real materials and simulations of model glass formers and to examine the generality of the correlation between fragility and dynamic heterogeneity observed here.

Although both  $\tau$  and  $\tau_{\alpha_{2,\text{max}}}$  increase upon cooling, they are usually not linearly related; i.e., the characteristic times for structural relaxation and establishment of  $\alpha_{2,\text{max}}$  decouple [25,37]. As shown in Fig. S4 of the Supplemental Material [31], the  $\tau$  versus  $\tau_{\alpha_{2,\text{max}}}$  curves for systems with different fragilities deviate a lot. A scaling collapse of  $\tau(\tau_{\alpha_{2,\text{max}}})$  has been achieved by simply adjusting a systemdependent scaling factor to rescale  $\tau_{\alpha_{2,\text{max}}}$  [25]. However, the physical meaning of the manipulative scaling factor is unclear. Interestingly, when we plot  $\tau/\tau^*$  against  $\tau_{\alpha_{2,\text{max}}}/\tau^*$ , as shown in Fig. 2(c), curves for all systems studied collapse onto the same master curve,

$$\tau/\tau^* = H(\tau_{\alpha_{2\max}}/\tau^*),\tag{3}$$

where *H* is also piecewise. Since  $\tau^*$  is intrinsically equivalent to *m*, the decoupling relation between  $\tau$  and  $\tau_{\alpha_{2,\max}}$  is controlled as well by the kinetic fragility, which is another robust evidence confirming that the kinetic fragility is the key to connecting structural relaxation and dynamic heterogeneity.

Now we have seen the essential role of the kinetic fragility or the characteristic time scale in unifying the relationship between structural relaxation and dynamic heterogeneity. This further stimulates our ambition to find a general description of  $\tau(T)$ . Note that  $\tau^*$  hides behind an identical  $\alpha_{2,\text{max}}$ , which couples with a system-dependent temperature  $T^*$ . Now that we have shown the importance of  $\tau^*$ , it is interesting to know whether  $T^*$  is crucial as well.

Similar to what has been done for Fig. 2(a), we shift all  $\alpha_{2,\max}(T)$  curves (see examples in Fig. S5 in the Supplemental Material [31]) to that of the Hertzian one at  $P = 5.00 \times 10^{-3}$  and take the Hertzian state at  $T = T^* = 1.46 \times 10^{-4}$  and  $P = 5.00 \times 10^{-3}$  as the reference. This leads to a nice scaling collapse,

$$\alpha_{2,\max} = f_T(T^*/T),\tag{4}$$

as shown in Fig. 2(d). Equation (4) verifies that  $T^*$  is indeed the characteristic temperature we are looking for. Like  $f_{\tau}$  in Eq. (1) and H in Eq. (3),  $f_T$  in Eq. (4) is piecewise as well. Similarly, piecewise behaviors can also be observed when  $\chi_{4,\text{max}}$  (the maximum of  $\chi_4$ ) is plotted as a function of  $\tau$  or T[26,27,31]. The initial power-law and then a logarithmic growth of  $\chi_{4,\text{max}}$  with  $\tau$  can be predicted, respectively, by mode coupling and random first order transition theories, though details regarding the crossover between the two regimes of growth are still puzzling [14]. Since  $\alpha_{2,\text{max}}$ grows less strongly than  $\chi_{4,\text{max}}$  with  $\tau$  or T (see Fig. S10 in the Supplemental Material [31]), it may be interesting to check whether theories that can predict behaviors of  $\chi_{4,\text{max}}(\tau)$  are also applicable to  $\alpha_{2,\text{max}}(\tau)$ .

Unlike that  $\tau^*$  has a one-to-one correspondence with m, we find no direct correlation between  $T^*$  and m; e.g., two systems with LJ 12-6 potential at P = 6 and P = 10, respectively, have equal m but pretty different values of  $T^*$  [31]. Equation (4) hints that, although  $T^*$  is sensitive to system parameters (interaction potentials, pressures, etc.), it may be coupled to other characteristic temperatures, e.g., the glass transition temperature, which is crucial to establish the general description of  $\tau(T)$  for various systems shown in the following.

Now we are going to move one step further to discuss the scaling collapse of  $\tau(T)$ . To unify in the same framework different dynamic slowdown in various glass-forming liquids, people have tried to manipulate the scaling collapse of dynamics in different ways. An excellent scaling collapse of  $\tau(T)$  in Lennard-Jones systems has been achieved by using a density scaling function [8]. However, the density scaling procedure usually yields different scaling curves for different systems and fails in some systems, e.g., systems with harmonic potentials studied here. Moreover, it has been shown that existing methods to achieve scaling collapse of  $\tau(T)$  for specific systems cannot be simply generalized to other systems [9–11]. Interestingly, the combination of Eqs. (1) and (4) can lead to the general scaling relation of  $\tau(T)$ ,

$$\tau/\tau^* = f_\tau^{-1}[f_T(T^*/T)] = F(T^*/T).$$
(5)

Therefore, by introducing  $\tau^*$  and  $T^*$ , the long-sought scaling collapse of  $\tau(T)$  for various systems is straightforward, as corroborated in Fig. 3. To our knowledge, so far, there has been no work to successfully collapse  $\tau(T)$  for so many systems with vastly different potentials and over so wide a range of pressures and fragilities without introducing additional or arbitrary parameters. The scaling collapse shown in Fig. 3 only involves characteristic scales associated with an identical dynamic heterogeneity, which have clear physical meanings. Dynamic heterogeneity is believed to be important in understanding the glass transition, which is directly and confirmatively evidenced here by the scaling collapses shown in Figs. 2 and 3.

Next, we study the functional form for F(x) in Eq. (5). As mentioned earlier, there are multiple functions proposed to fit  $\tau(T)$ . In the inset to Fig. 3, we show that VFT, MC, ECG, AM, and MYEGA forms can all fit  $\tau(T)$  well for a single system. However, when we try to fit the master curve in the main panel of Fig. 3 using these five forms, only VFT can fit the whole curve nicely, while the other four forms can only fit the high  $T^*/T$  part well, which mainly contains more fragile liquids within the simulation time window. Though the VFT form can describe our master curve well, it should also be noted that the VFT description of  $\tau(T)$  is



FIG. 3. Scaled structural relaxation time  $\tau/\tau^*$  versus scaled reciprocal temperature  $T^*/T$  for all systems studied. Black solid curve is the VFT fit:  $y = 0.00337 \exp[1.606/(x - 0.705)]$ , where  $x = T/T^*$  and  $y = \tau/\tau^*$ . Red dashed curve is a fit to the MC form:  $y = 0.00314(x - 0.809)^{-3.441}$ . Blue dashed-dotted curve indicates the ECG fit:  $y = 0.373 \exp[95.845(x^{-1} - 0.917)^2]$ . Navy dashed-dotted curve is a fit to the AM form:  $y = 0.00943 \exp[(1.284/x)^{5.542}]$ . Magenta dotted curve is the MYEGA fit:  $y = 0.00782 \exp[(0.0730/x) \exp(4.052/x)]$ . (Inset)  $\tau(T)$  for a Hertz system at  $P = 5 \times 10^{-7}$ , whose corresponding scaled data lie in the shadowed region in the main panel. Note that the curve in the inset can be fitted well with all the above five forms before scaling. After scaling, it lies in the region where only VFT works in the main panel.

challenged in an experimental study [45], where ultraviscous molecular liquids were studied.

Our numerical studies of different model glass formers unravel a general description of the dynamics during vitrification. For the underlying connection between dynamic heterogeneity and structural relaxation, the key is the awareness of the importance of the constant dynamic heterogeneity condition and the characteristic scales hiding behind it. By introducing the characteristic time scale and temperature under the iso- $\alpha_{2,max}$  condition, scaling collapses regarding the structural relaxation and dynamic heterogeneity can be generally described. It reveals the longsought general description of the relationship between structural relaxation time and temperature without introducing any adjustable parameter in various glass formers. Since the characteristic time scale is equivalent to kinetic fragility, it is suggested that the kinetic fragility serves as the link between structural relaxation and dynamic heterogeneity. Moreover, our major conclusions hold as well if dynamic heterogeneity is quantified by dynamic susceptibility, as discussed systematically in the Supplemental Material [31]. Our work suggests that dynamic heterogeneity plays a more important role than expected in studying the nature of the glass transition.

Though our major findings do not rely on whether dynamic heterogeneity is measured by  $\alpha_{2,\text{max}}$  or  $\chi_{4,\text{max}}$ , it should be noted that  $\alpha_{2,\text{max}}$  and  $\chi_{4,\text{max}}$  (see comparison between  $\alpha_{2,\text{max}}$  and  $\chi_{4,\text{max}}$  in Fig. S10 of the Supplemental Material [31]), as well as the size of cooperatively rearranging regions proposed in the Adam-Gibbs model [46], are only qualitatively equivalent measures of dynamic heterogeneity, because quantitative inconsistencies of their temperature dependence can be observed [47]. This can also to some extent be implied by the observation that  $\tau_{\alpha_{2,\text{max}}}(\tau)$  is piecewise, while  $\tau_{\chi_{4,\text{max}}} \sim \tau$  [15,23] [see Figs. S4 and S6(b) of the Supplemental Material [31]] with  $\tau_{\chi_{4,\text{max}}}$ being the time when  $\chi_{4,\text{max}}$  occurs.

Our findings here are based on numerical studies of molecular glass formers, and the glass transition temperature defined here is higher than experiment [3,27,48], which thus calls for further experimental verification of our findings. To our knowledge, the development of an estimate of  $\chi_{4,\text{max}}$  in Ref. [15] has advanced greatly the experimental studies of dynamic heterogeneity in real materials. Since  $\alpha_{2,max}$  is also important in measuring dynamic heterogeneity from our study, it will be meaningful to devise an experimentally measurable estimate of  $\alpha_{2,\max}$ . Probing dynamic heterogeneity precisely is challenging in experiments of molecular glass formers [14,15,20,22,27], while it is feasible in experiments of colloids [49–51]. Recent studies [10,49] have demonstrated that some behaviors of colloidal and molecular glass formers show remarkable similarities, and hence it is intriguing to see whether the scenarios reported here can also be observed in colloidal experiments.

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- [1] P.G. Debenedetti and F.H. Stillinger, Nature **410**, 259 (2001).
- [2] H. Tanaka, T. Kawasaki, H. Shintani, and K. Watanabe, Nat. Mater. 9, 324 (2010).
- [3] L. Berthier and G. Biroli, Rev. Mod. Phys. 83, 587 (2011).
- [4] Y. S. Elmatad, D. Chandler, and J. P. Garrahan, J. Phys. Chem. B 113, 5563 (2009).
- [5] I. Avramov and A. Milchev, J. Non-Cryst. Solids 104, 253 (1988); I. Avramov, J. Non-Cryst. Solids 238, 6 (1998).
- [6] J. C. Mauro, Y. Yue, A. J. Ellison, P. K. Gupta, and D. C. Allan, Proc. Natl. Acad. Sci. U.S.A. 106, 19780 (2009).
- [7] C. A. Angell, J. Non-Cryst. Solids 131–133, 13 (1991).
- [8] L. Bøhling, T. S. Ingebrigtsen, A. Grzybowski, M. Paluch, J. C. Dyre, and T. B. Schrøder, New J. Phys. 14, 113035 (2012).
- [9] P. E. Ramírez-González, L. López-Flores, H. Acuna-Campa, and M. Medina-Noyola, Phys. Rev. Lett. 107, 155701 (2011).
- [10] N. Xu, T. K. Haxton, A. J. Liu, and S. R. Nagel, Phys. Rev. Lett. 103, 245701 (2009).
- [11] L. Berthier and G. Tarjus, J. Chem. Phys. 134, 214503 (2011).
- [12] W. Kob, C. Donati, S. J. Plimpton, P. H. Poole, and S. C. Glotzer, Phys. Rev. Lett. 79, 2827 (1997).
- [13] M. D. Ediger, Annu. Rev. Phys. Chem. 51, 99 (2000).
- [14] Dynamical Heterogeneities in Glasses, Colloids, and Granular Media, edited by L. Berthier, G. Biroli, J.-P. Bouchaud, L. Cipelletti, and W. van Saarloos (Oxford University Press, New York, 2011).
- [15] L. Berthier et al., Science 310, 1797 (2005).
- [16] E. Flenner and G. Szamel, Phys. Rev. Lett. 105, 217801 (2010).
- [17] N. Lacevic, F. W. Starr, T. B. Schrøder, and S. C. Glotzer, J. Chem. Phys. **119**, 7372 (2003).
- [18] E. Flenner, H. Staley, and G. Szamel, Phys. Rev. Lett. 112, 097801 (2014).
- [19] D. Coslovich and C. M. Roland, J. Non-Cryst. Solids 357, 397 (2011).
- [20] K. Koperwas, A. Grzybowski, K. Grzybowska, Z. Wojnarowska, A. P. Sokolov, and M. Paluch, Phys. Rev. Lett. 111, 125701 (2013).
- [21] R. Casalini, D. Fragiadakis, and C. M. Roland, J. Chem. Phys. 142, 064504 (2015).
- [22] D. Fragiadakis, R. Casalini, and C. M. Roland, J. Phys. Chem. B 113, 13134 (2009).

- [23] D. Coslovich and C. M. Roland, J. Chem. Phys. 131, 151103 (2009).
- [24] A. Grzybowski, K. Koperwas, K. Kolodziejczyk, K. Grzybowska, and M. Paluch, J. Phys. Chem. Lett. 4, 4273 (2013).
- [25] P. Henritzi, A. Bormuth, F. Klameth, and M. Vogel, J. Chem. Phys. 143, 164502 (2015).
- [26] L. Berthier, Physics 4, 42 (2011).
- [27] C. Dalle-Ferrier, C. Thibierge, C. Alba-Simionesco, L. Berthier, G. Biroli, J.-P. Bouchaud, F. Ladieu, D. L'Hôte, and G. Tarjus, Phys. Rev. E 76, 041510 (2007).
- [28] A. Grzybowski, K. Kolodziejczyk, K. Koperwas, K. Grzybowska, and M. Paluch, Phys. Rev. B 85, 220201(R) (2012).
- [29] L. Wang, Y. Duan, and N. Xu, Soft Matter 8, 11831 (2012).
- [30] S. E. Abraham, S. M. Bhattacharrya, and B. Bagchi, Phys. Rev. Lett. **100**, 167801 (2008).
- [31] See Supplemental Material at http://link.aps.org/ supplemental/10.1103/PhysRevLett.120.125502 for more details about our numerical models, as well as values of m,  $\tau^*$ , and  $T^*$ , and for further results from the four-point dynamic susceptibility, which includes Refs. [32–34].
- [32] P. Bordat, F. Affouard, M. Descamps, and K. L. Ngai, Phys. Rev. Lett. 93, 105502 (2004).
- [33] Y. Q. Cheng, E. Ma, and H. W. Sheng, Phys. Rev. Lett. 102, 245501 (2009).
- [34] S. Plimpton, J. Comput. Phys. 117, 1 (1995).
- [35] W. Kob and H. C. Andersen, Phys. Rev. Lett. 73, 1376 (1994).
- [36] Computer Simulation of Liquids, edited by M. P. Allen and D. J. Tildesley (Oxford Science Publications, New York, 1987).
- [37] F. W. Starr, J. F. Douglas, and S. Sastry, J. Chem. Phys. 138, 12A541 (2013).
- [38] K. Kim and S. Saito, J. Chem. Phys. 138, 12A506 (2013).
- [39] L. Berthier and T. A. Witten, Europhys. Lett. 86, 10001 (2009).
- [40] L. Wang, P. Guan, and W. H. Wang, J. Chem. Phys. 145, 034505 (2016).
- [41] M. Paluch, E. Masiewicz, A. Grzybowski, S. Pawlus, J. Pionteck, and Z. Wojnarowska, J. Chem. Phys. 141, 134507 (2014).
- [42] K. Grzybowska, S. Pawlus, M. Mierzwa, M. Paluch, and K. L. Ngai, J. Chem. Phys. **125**, 144507 (2006); J. Phys. Chem. Lett. **1**, 1170 (2010).
- [43] R. Casalini and C. M. Roland, Phys. Rev. B 71, 014210 (2005).
- [44] D. Fragiadakis, R. Casalini, and C. M. Roland, Phys. Rev. E 84, 042501 (2011).
- [45] T. Hecksher, A. I. Nielsen, N. B. Olsen, and J. C. Dyre, Nat. Phys. 4, 737 (2008).
- [46] G. Adam and J. H. Gibbs, J. Chem. Phys. 43, 139 (1965).
- [47] X. Di and G. B. McKenna, J. Chem. Phys. 138, 12A530 (2013).
- [48] A. Ninarello, L. Berthier, and D. Coslovich, Phys. Rev. X 7, 021039 (2017).
- [49] J. Mattsson, H. M. Wyss, A. Fernandez-Nieves, K. Miyazaki, Z. Hu, D. R. Reichman, and D. A. Weitz, Nature 462, 83 (2009).
- [50] E. R. Weeks et al., Science 287, 627 (2000).
- [51] S. Goldel, T. Palberg, and H. Joachim Schöpe, Nat. Phys. 12, 712 (2016).