

ARTICLE

https://doi.org/10.1038/s41467-022-30454-w

OPEN



Superconductivity above 200 K discovered in superhydrides of calcium

Zhiwen Li^{1,2,9}, Xin He^{1,2,3,9}, Changling Zhang^{1,2,9}, Xiancheng Wang ^{1,2 ×}, Sijia Zhang¹, Yating Jia¹, Shaomin Feng¹, Ke Lu^{1,2}, Jianfa Zhao^{1,2}, Jun Zhang ^{1,2}, Baosen Min^{1,2}, Youwen Long ^{1,2,3}, Richeng Yu^{1,2}, Luhong Wang⁴, Meiyan Ye⁵, Zhanshuo Zhang⁵, Vitali Prakapenka⁶, Stella Chariton⁶, Paul A. Ginsberg⁷, Jay Bass⁷, Shuhua Yuan⁸, Haozhe Liu ⁸ & Changqing Jin ^{1,2,3 ×}

Searching for superconductivity with T_c near room temperature is of great interest both for fundamental science & many potential applications. Here we report the experimental discovery of superconductivity with maximum critical temperature (T_c) above 210 K in calcium superhydrides, the new alkali earth hydrides experimentally showing superconductivity above 200 K in addition to sulfur hydride & rare-earth hydride system. The materials are synthesized at the synergetic conditions of 160~190 GPa and ~2000 K using diamond anvil cell combined with in-situ laser heating technique. The superconductivity was studied through in-situ high pressure electric conductance measurements in an applied magnetic field for the sample quenched from high temperature while maintained at high pressures. The upper critical field Hc(0) was estimated to be ~268 T while the GL coherent length is ~11 Å. The in-situ synchrotron X-ray diffraction measurements suggest that the synthesized calcium hydrides are primarily composed of CaH₆ while there may also exist other calcium hydrides with different hydrogen contents.

¹ Beijing National Laboratory for Condensed Matter Physics, Institute of Physics, Chinese Academy of Sciences, 100190 Beijing, China. ² School of Physical Sciences, University of Chinese Academy of Sciences, 100190 Beijing, China. ³ Songshan Lake Materials Laboratory, 523808 Dongguan, China. ⁴ Harbin Institute of Technology, 130001 Harbin, China. ⁵ Jilin University, Changchun, China. ⁶ Center for Advanced Radiations Sources, University of Chicago, Chicago, IL 60637, USA. ⁷ Department of Geology, University of Illinois at Urbana-Champaign, Urbana, IL 61801, USA. ⁸ Center for High Pressure Science & Technology Advanced Research, 100094 Beijing, China. ⁹ These authors contributed equally: Zhiwen Li, Xin He, Changling Zhang. [⊠]email: Wangxiancheng@iphy.ac.cr; Jin@iphy.ac.cr

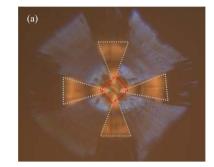
uperhydrides have received growing attentions recently because of high temperature superconductivity (SC)¹. It was suggested that superhydrides are likely to be a step toward metallic hydrogen since the "chemically precompressed" hydrogen in the superhydrides can decrease the critical pressure required for the metallization¹⁻³. Because of high phonon frequencies and strong electron-phonon coupling, high temperature SC was suggested in hydrogen rich materials. Sulfur hydride is one of example in which the high T_c SC was reported both theoretically & experimentally at megabar level pressures^{4,5}. In addition, alkali-earth, rare-earth, and other superhydrides are theoretically suggested to possibly possess high temperature SC at high pressures⁶⁻¹¹. Those metal superhydrides favor to form hydrogen rich coordination such as clathrate-like cage structures at high pressure where the metal ions are located in the center of the H clathrates and act as carrier donor. The occupation of the unfilled anti-bonding σ^* orbitals of the H₂ molecules by the electrons from the metal ions would enhance to dissociate the H₂ molecules hence form H clathrates. Since the discovery of SC in sulfur hydrides, other superhydrides have been experimentally reported to have SC, including LaH10¹¹⁻¹⁴, YH9^{8,15}, CeH9¹⁶, ThH₁₀¹⁷, zirconium polyhydrides¹⁸, tin hydrides¹⁹, YH₄²⁰, and LuH_3^{21} etc.

Although alkali-earth superhydrides were predicted to have high temperature SC, so far there are no experimental observations for SC above 200 K, in spite of SC with $T_c \sim 20$ K was reported in barium hydrides²². Here, we report the experimental discovery of high T_c SC in calcium superhydrides synthesized at high pressure and high temperature conditions. The SC with $T_c \sim 210$ K at 160 ~190 GPa was observed. Hence in addition to sulfur hydrides and rare-earth hydrides, this is a new alkali-earth hydride superconductor with T_c above 200 K.

Results and discussions

High-temperature superconductivity. Figure 1a displays the image of the sample assembly. The dotted red square is the sample shape and the four gray dotted regions are the inner electrodes composed of Pt. Figure 1b shows the measurements of resistance as function of temperature at 160 GPa for Sample A. The resistance curves almost coincide for both the cooling and warming cycles. The resistance drops at around 210K and approaches to zero at low temperature, suggesting that the superconducting like transitions take place in the calcium superhydrides. Figure 2 shows the resistance measurements by application of magnetic fields for Sample A. It is seen that the whole transition curves shift to low temperature with the critical temperature gradually suppressed by the applied field, consistent with the nature of superconductivity. The inset of Fig. 1b displays an enlarged view of the derivative of resistance over temperature around the transition region, from which the $T_{\rm c}^{\rm onset}$ of 210 K is estimated referring to right side upturn point. The experimentally observed T_c value of 210 K is comparable to the theoretical predictions for the calcium hydride CaH₆ that crystalizes in a cubic structure with the space group of Im-3m⁶.

In addition to the first transition at ~210 K, there are also several drop kinks at ~180 K and ~160 K as shown in Fig. 1b. These drop kinks imply multistep superconducting transitions that are supported by the broaden transition curves shown in Fig. 2. In the context of theoretical calculation, the calcium hydrides such as CaH₆, CaH₁₂ etc are predicted to have high T_c SC at high pressure^{2,6}. Since near the experimental pressure of 160 GPa, CaH₄, CaH₆, and CaH₁₂ are predicted to be stable, the stoichiometric ratio of the synthesized superconducting hydrides CaHx should be mainly determined by the hydrogen content



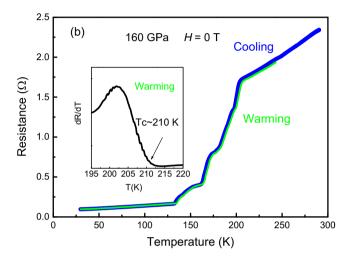


Fig. 1 The superconductivity measurements. a The image of the specimen assembly. **b** The temperature dependence of resistance R(T) measured at 160 GPa and zero magnetic field for Sample A. The multiple steps in the transition region imply there might be phases with different hydrogen amount in the synthesized calcium superhydrides. The inset is the enlarged view of the derivative of resistance over temperature(dR/dT) where the ~210 K T_c^{onset} is determined referring to the upturn point at the right side.

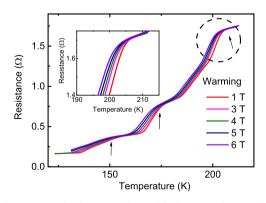


Fig. 2 The superconducting transitions with the applied magnetic fields. The resistance as a function of temperature measured at different magnetic fields for Sample A. The arrows are attributed to the phases with different hydrogen amount in the calcium superhydrides. The inset is the enlarged view of the resistance curves around the transition region of the 210 K phase.

released during the laser heating process. Hence it is likely that these multistep transitions should be attributed to calcium hydrides with different hydrogen amounts, which are probably due to inhomogeneous hydrogen distribution.

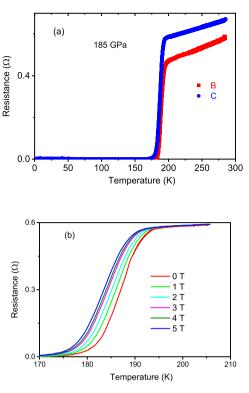


Fig. 3 Evolution of superconducting transitions as functions of magnetic fields. a The temperature dependence of resistance measured for Samples B and C at 185 GPa showing one-step transition; **b** Resistance curves measured under different magnetic field for Sample C.

To increase homogeneity of the generated hydrogen, more homogenous scanning heating manner controlled by computer was adopted. We performed several sample synthesis to achieve a single superconducting phase transition by optimizing pressure and temperature conditions. Figure 3a presents the resistance for Samples B and C as function of temperature measured at 185 GPa, where a sharp superconducting transition is achieved. The SC is supported by the measurements at applied magnetic field as shown in Fig. 3b. For the Sample D, the superconducting transition is shown in Supplementary Fig. 1. Thus, our results show the experimental discovery of high temperature superconducting alkali earth superhydride with $T_{\rm c}$ exceeding 200 K, joining the 200 K above superhydrides of SH₃ and LaH₁₀ and YH₉. Since hydrogen is the lightest element, superhydrides are expected to have high Debye temperature and thus possible high T_c SC according to the prediction by conventional BCS theory. In addition, for the reported binary superhydrides with high T_c SC, it was found that most of elemental metals are on the boundary of s and d blocks in the elemental period table², such as La in LaH₁₀ and Y in YH₉. For these elemental metals, electrons can be transferred from filled s orbital to d orbital under high pressure due to the small energy difference between these orbitals, which usually increases the structural instability and enhances the electron-phonon coupling strength and thus leads to possible SC with high T_c as well^{2,23,24}. The Ca lattice in CaH₆ predicted in ref. ⁶ is highly compressed under high pressure, which suggests the possible large electron phonon coupling strength caused by the electron s-d orbital transfer. Therefore, the combination of the high Debye temperature and the large electron-phonon coupling strength makes it possible to reach high T_c SC in calcium superhydrides.

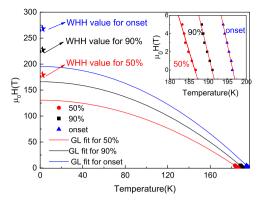


Fig. 4 The superconducting parameters. The Ginzburg Landau fitting for the $H_{c2}(T)$ shown with the solid lines. The stars represent the $H_{c2}(0)$ values calculated via WHH model. The inset of Fig. 4 is the critical field Hc2(T) as a function of T_c with the T_c values determined by the criteria of the T_c^{onset} , 90 and 50% of normal state resistance, respectively.

Physics parameters measurements. To estimate the upper critical field $H_{c2}(0)$, the superconductivity as a function of magnetic field was investigated based on Sample C. The T_c values are determined with the criteria of the T_{c}^{onset} , the temperature at 90 and 50% of normal state resistance ($T_c^{90\%}$ and $T_c^{50\%}$), respectively. By linear fitting the H_{c2} (T), the slope of $dH_{c2}/dT|_{Tc}$ was obtained to be -1.976 T/K for T_c^{onset} , -1.726 T/K for $T_c^{90\%}$, and -1.389 T/K for the $T_c^{50\%}$. Using the Werthamer-Helfand-Hohenberg (WHH) formula of $\mu_0 H_{c2}(T) = -0.69 \times dH_{c2}/dT|_{T_c} \times T_c$, the $H_{c2}(0)$ value was calculated to be 268T, 226T, and 180T for the criteria of T_c^{onset} , $T_c^{90\%}$, and $T_c^{50\%}$, respectively. The Ginzburg Landau (GL) formula of $\mu_0 H_{c2}(T) = \mu_0 H_{c2}(0)(1 - (T/T_c)^2)$ is also used to estimate the upper critical field at zero temperature. As shown in Fig. 4 the fitting of the $\mu_0 H_{c2}(T)$ by GL formula gives a value of $H_{c2}(0)$ ranging from 131T to 196T using criteria of T_c^{onset} , $T_c^{90\%}$ and $T_c^{50\%}$, respectively. These are comparable to values reported for LaH_{10}^{12} . Using the obtained value of $H_{c2}(0) = 131-268 \text{ T}$, one can roughly estimate the GL coherence length ξ to be 11–16 Å via the equation of $\mu_0 H_{c2}(0) = \Phi_0 / 2\pi \xi^2$, where $\Phi_0 = 2.067 \times 10^{-15}$ Web is the magnetic flux quantum.

Structure characterizations. The possible superconducting phases are investigated by high pressure X-ray diffraction experiments. Supplementary Fig. 2a shows the room temperature X-ray diffraction pattern for a synthesized sample, where the $Im\bar{3}m$ phase of CaH₆ was indexed. Supplementary Figure 2b shows the X-ray diffraction patterns measured at various high pressures starting from the synthesized pressure up to 243 GPa where the anvils were broken. The pressure dependence of unit cell volume of the CaH₆ is presented in Fig. 5. By using the Birch–Murnaghan Eq. (1),

$$P(\text{GPa}) = \frac{3}{2} \times B_0 \left[\left(\frac{V_0}{V} \right)^{\frac{2}{3}} - \left(\frac{V_0}{V} \right)^{\frac{5}{3}} \right] \\ \times \left\{ 1 - \left(3 - \frac{3}{4} \times B'_0 \right) \times \left[\left(\frac{V_0}{V} \right)^{\frac{2}{3}} - 1 \right] \right\}$$
(1)

where B_0 is the bulk modulus and B_0 ' is the pressure derivative bulk modulus, the *P*-*V* data are fitted and B_0 is estimated to be about 221 GPa at $B_0' = 3$. From the fit of equation of state the lattice constant *a* at 150 GPa is estimated to be 3.308 Å, which is close to that predicted by theoretical calculation. This suggests that the superconducting sample contains $Im\bar{3}m$ phase of

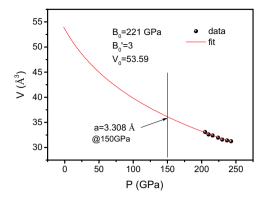


Fig. 5 The structure characterizations. The pressure dependence of unit cell volume of the CaH_6 . The red line is the fit to equation of stat.

 CaH_6 . However, other phase(s) might be mixed since there are some unknown weak peaks in the diffraction patterns. We became aware during preparing the paper that an independent work by Ma et al. was carried with the similar results²⁵.

In summary high Tc calcium superhydrides superconductors are discovered with critical temperature Tc above 210 K at pressures ~160 GPa. The upper critical field Hc(0) was estimated to be ~268T that corresponds to ~11 Å coherent length in GL model for the calcium superhydride superconductor. The results show that calcium superhydrides are new type hydrogen rich superconducting compounds with transition temperature above 200 K in addition to sulfur hydride & rare-earth hydride system.

Methods

High pressure assembly. Diamond anvil cell techniques are used for both synthesis and measurements of the calcium hydrides. The diamond anvils are of optical pure quality with culet diameter of 100 µm beveled to 300 µm for the experiments. The electrodes for resistance measurements are embedded in the sample chamber before high pressure and high temperature synthesis. The experiments are conducted in two steps: (I) synthesis of the calcium hydrides at high pressure and high temperature; (II) the follow up measurements of SC for the sample quenched from high temperature but remained at the same pressure. The T301 stainless steel served as gaskets that are prepressed from $250 \,\mu\text{m}$ to $\sim 20 \,\mu\text{m}$ in thickness at $\sim 20 \,\text{GPa}$. A hole of 300 µm in diameter was drilled at the center of imprint, and was filled with fine insulating cubic boron nitride (cBN) powder mixed with epoxy that was pressed with anvils to form a solid layer of ~15 μ m in thickness. A hole of 70 μ m in diameter was drilled at the center of the solid cBN layer to serve as the sample chamber. The Pt foils with a thickness of 0.5 µm were deposited on the surface of the culet as the inner electrodes. Gold wire was attached to the Pt foil to serve as the outside electrodes. High purity calcium metal (99.9%) was adopted as precursor material: a 2-µm thick calcium specimen with dimensions $30 \,\mu\text{m} \times 30 \,\mu\text{m}$ was stacked on the inner electrodes. A flake of ammonia borane was inserted into the sample chamber to act as a hydrogen source (hydrodizer) while also serving as the pressure transmitting medium¹⁴. The sample loading as well as electrodes deposition were conducted in a glove box filled by flowing Ar gas with 1 ppm less trace water or oxygen to avoid moisture or contamination. The diamond anvil cell was then clamped in the glove box before the measurements. The described setup followed the ATHENA procedure reported ref. 26.

Sample synthesis and resistance measurements. A piston cylinder type anvil cell composed of BeCu was adopted in the experiments. The sample was first pressed to 160–190 GPa followed by heating at ~2000 K for ~6 min. The heating was performed with a pulse laser beam of 1064 nm wavelength. The laser had a focused spot size 5 μ m in diameter with a power of 20 W. The heating temperature was estimated by fitting the black body irradiation spectra. A computer controlled scanning heating method was adopted to increase the homogeneity of the generated hydrogen. The heated sample was then quenched from high temperature while held at high pressure. The synthesized sample compressed within diamond anvil cell was put into a MagLab system to perform the electric conductivity measurements using a Van der Pauw method with 1 mA applied current. The MagLab system can provide synergetic extreme environments with temperatures from 300 K to 1.5 K and a magnetic field up to 9T^{27,28}. Pressure was calibrated by the shift of the first order Raman edge frequency from the cutlet of diamond²⁹.

High pressure x-ray diffraction measurements. The synchrotron X-ray diffraction experiments in the diamond anvil cell are carried out at GSECARS of the Advanced

Photon Source at the Argonne National Laboratory. X-ray with $\lambda=0.3344$ Å are focused down to ~3 μm diameter spot on the sample. A symmetric diamond anvil cell was used to generate pressure with beveled diamond anvils (100/300 μm) and a rhenium gasket indented to 25 μm . Samples are loaded into the pressure chamber without cBN insulating layer. Calcium hydrides are synthesized in situ at 190 GPa with laser heating at 2000 K. The sample pressure in the synchrotron X-ray diffraction experiments was determined by both the shift of Raman edge frequency from the cutlet of diamond and the equation of state for rhenium. The X-ray diffraction images are converted to one dimensional diffraction data with Dioptas³⁰.

Data availability

All the data generated in the study are available upon reasonable request to the corresponding author.

Received: 30 November 2021; Accepted: 27 April 2022; Published online: 23 May 2022

References

- Ashcroft, N. W. Hydrogen dominant metallic alloys: high temperature superconductors? *Phys. Rev. Lett.* 92, 187002 (2004).
- Semenok, D. V. et al. On distribution of superconductivity in metal hydrides. Curr. Opin. Solid State Mater. Sci. 24, 100808 (2020).
- 3. Xu, J. A. & Zhu, Z. W. Metallic hydrogen. Physics 6, 296 (1977).
- Duan, D. F. et al. Pressure-induced metallization of dense (H2S)(2)H-2 with high-T-c superconductivity. Sci. Rep. 4, 6968 (2014).
- Drozdov, A. P. et al. Conventional superconductivity at 203 kelvin at high pressures in the sulfur hydride system. *Nature* 525, 73 (2015).
- Wang, H. et al. Superconductive sodalite-like clathrate calcium hydride at high pressures. Proc. Natl Acad. Sci. USA 109, 6463 (2012).
- 7. Peng, F. et al. Hydrogen clathrate structures in rare earth hydrides at high pressures: possible route to room-temperature superconductivity. *Phys. Rev. Lett.* **119**, 107001 (2017).
- Kong, P. P. et al. Superconductivity up to 243 K in the yttrium-hydrogen system under high pressure. *Nat. Commun.* 12, 5075 (2021).
- Liu, H. Y. et al. Potential high-T-c superconducting lanthanum and yttrium hydrides at high pressure. *Proc. Natl Acad. Sci. USA* 114, 6990 (2017).
- Flores-Livas, J. A. et al. A perspective on conventional high-temperature superconductors at high pressure: Methods and materials. *Phys. Rep.* 856, 1–78 (2020).
- Geballe, Z. M. et al. Synthesis and stability of lanthanum superhydrides. Angew. Chem. Int. Ed. 57, 688–692 (2018).
- Drozdov, A. P. et al. Superconductivity at 250 K in lanthanum hydride under high pressures. *Nature* 569, 528 (2019).
- Hong, F. et al. Superconductivity of lanthanum superhydride investigated using the standard four-probe configuration under high pressures. *Chin. Phys. Lett.* 37, 107401 (2020).
- 14. Somayazulu, M. et al. Evidence for superconductivity above 260 K in lanthanum superhydride at megabar pressures. *Phys. Rev. Lett.* **122**, 027001 (2019).
- Snider, E. et al. Synthesis of yttrium superhydride superconductor with a transition temperature up to 262 K by catalytic hydrogenation at high pressures. *Phys. Rev. Lett.* **126**, 117003 (2021).
- Chen, W. H. et al. High-temperature superconducting phases in cerium superhydride with a T-c up to 115 K below a pressure of 1 megabar. *Phys. Rev. Lett.* 127, 117001 (2021).
- Semenok, D. V. et al. Superconductivity at 161 K in thorium hydride ThH10: Synthesis and properties. *Mater. Today* 33, 36–44 (2020).
- Zhang, C. L. et al. Superconductivity in zirconium polyhydrides with Tc above 70K. Sci. Bull. 67, 907 (2022).
- Hong, F. et al. Possible superconductivity at ~70 K in tin hydride SnHx under high pressure. *Mater. Today Phys.* 22, 100596 (2022).
- Shao, M. Y. et al. High-pressure synthesis of superconducting clathratelike YH4. *Phys. Rev. B* 104, 174509 (2021).
- Shao, M. Y. et al. Superconducting ScH3 and LuH3 at megabar pressures. Inorg. Chem. 60, 15330–15335 (2021).
- 22. Chen, W. et al. Synthesis of molecular metallic barium superhydride: pseudocubic BaH12. *Nat. Commun.* **12**, 273 (2021).
- Maksimov, E. G., Magnitskaya, M. V. & Fortov, V. E. Non-simple behavior of simple metals at high pressure. *Phys. -Uspekhi* 48, 761–780 (2005).
- Semenok, D. V., Kvashnin, A. G., Kruglov, I. A. & Oganov, A. R. Actinium hydrides AcH10, AcH12, and AcH16 as high-temperature conventional superconductors. J. Phys. Chem. Lett. 9, 1920–1926 (2018).
- Ma, L. et al. Experimental observation of superconductivity at 215 K in calcium superhydride under high pressures. arXiv https://doi.org/10.48550/ arXiv.2103.16282 (2021).

- 26. Jia, Y. T. et al. A combinatory package for diamond anvil cell experiments. *Crystals* **10**, 1116 (2020).
- 27. Zhang, S. J. et al. Effect of pressure on the iron arsenide superconductor Li_xFeAs (x=0.8,1.0,1.1). *Phys. Rev. B* **80**, 014506 (2009).
- Zhang, J. L. et al. Pressure-induced superconductivity in topological parent compound Bi₂Te₃. Proc. Natl Acad. Sci. USA 108, 24–28 (2011).
- Mao, H. K., Xu, J. & Bell, P. M. Calibration of the ruby pressure gauge to 800-Kbar under quasi-hydrostatic conditions. J. Geophys. Res.-Solid Earth Planets 91, 4673 (1986).
- Prescher, C. & Prakapenka, V. B. DIOPTAS: a program for reduction of twodimensional X-ray diffraction data and data exploration. *High. Press. Res.* 35, 223–230 (2015).

Acknowledgements

The work is supported by NSF, MOST, and CAS of China through research projects. Portions of this work were performed at GeoSoilEnviroCARS (The University of Chicago, Sector 13), Advanced Photon Source (APS), Argonne National Laboratory. GeoSoilEnviroCARS is supported by the National Science Foundation–Earth Sciences (EAR–1634415) and Department of Energy-GeoSciences (DE-FG02-94ER14466). This research used resources of the Advanced Photon Source, a U.S. Department of Energy (DOE) Office of Science User Facility operated for the DOE Office of Science by Argonne National Laboratory under Contract No. DE-AC02-06CH11357. We are grateful to Prof. J.G. Cheng, J.P. Hu, and L. Yu for the discussions. We thank Prof. T. Xiang, B.G. Shen, and Z.X. Zhao for the consistent encouragements.

Author contributions

Research design, coordination & supervision: C.Q.J.; high-pressure synthesis and in situ resistance measurements: Z.W. L., X.H., C.L.Z., S.J.Z., Y.T.J., S.M.F., Y.W.L., X.C.W., J.Z., B.S.M., R.C.Y, J.F.Z., and C.Q.J; in situ synchrotron experiments: L.H.W., M.Y.Y., Z.S.Z., V.B.P., S.C., P.A.G., J.B., S.H.Y., and H.Z.L., manuscript writing: Z.W.L., X.C.W., and C.Q.J. All authors contributed to the discussions.

Competing interests

The authors declare no competing interests.

Additional information

Supplementary information The online version contains supplementary material available at https://doi.org/10.1038/s41467-022-30454-w.

Correspondence and requests for materials should be addressed to Xiancheng Wang or Changqing Jin.

Peer review information *Nature Communications* thanks Yoshikazu Mizuguchi, and the other, anonymous, reviewer(s) for their contribution to the peer review of this work.

Reprints and permission information is available at http://www.nature.com/reprints

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit http://creativecommons.org/ licenses/by/4.0/.

© The Author(s) 2022